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Study on Corrosion Inhibitor in Mild Steel by Various Habitual Plant Extract – Review

K.Ramya*¹, N.Muralimohan²

¹Department Of Civil Engineering, Construction Engineering And Management,(2014-2016), K.S.R College Of Engineering,

²Department Of Civil Engineering, K.S.R College Of Engineering, Tiruchengode -637215 , Tamil Nadu , India,

Abstract : The inhibitor potentials of discrete plant extract was exploration by studying the corrosion behaviour of mild steel immersed in 2 M H₂SO₄ solution containing varied concentration of plant extract. Inhibitor efficiency of acid extract of plant such as musa sapientium peel , Emblica officinalis leaves and Azadirachta indica leaves as corrosion inhibitor for mild steel. The corrosion inhibition of the extract was criticized by using weight loss method , gasometric method , thermometric method and surface analysis studies . The inhibitor efficiency increased with increase in inhibitor concentration and decreased with temperature suggesting the occurrence of physical absorption and the highest % IE was obtained. As a result of literary work about performance was reviewed to identify the factor affecting the performance of corrosion inhibitor. Inhibitor efficiency of the extract is found to vary with concentration , temperature , period of immersion and pH. Value of active energy of the inhibited corrosion reaction of mild steel are greater than the value obtained for the blank solution. The mild steel surface in the absence and presence of the inhibitor was analysed by scanning electron microscopy(SEM) and the protective film have been analysed by Atomic force microscopic (AFM).

Keywords : Mild Steel , Adsorption, Natural Inhibitor, Musa Sapientium Peel , Emblica Officinalist Leaves And Azadirachta Indica .

1. Introduction

Mild steel ,which contain small percentage of carbon and is strong and easily worked but not readily tempered or harden. Due surroundings it tend to corrode , and is affected by numbers of factors such as , metal composition , temperature , presence of gases like sulphur dioxide , hydrogen chloride , chlorine and other corrosive gases , for structure . Few studies have conducted on the consistently available corrosion inhibitor potentials of some plant extract, which may know as “green inhibitor”. In the chemical environment , to prevent the corrosion rate of metal by using the green corrosion inhibitor. The adsorption of inhibitor can be enhanced by the presence of hetero atoms such as N, O, P and S. Corrosion processes develop fast after disruption of the protective barrier and are accompanied by a number of reactions that change the composition and properties of both the metal surface and the local environment, for example, formation of oxides, diffusion of metal into coating matrix, local pH. change and electrochemical potential. The study of corrosion of mild steel is a matter of tremendous theoretical and practical concern and as such as has received a considerable amount of interest. Acid solutions , widely used in industrial acid cleaning , acid descaling , acid pickling and oil well acidizing , require the use of corrosion inhibitors in order to restrain their corrosion attack on metallic materials.

In the view of various green corrosion inhibitions, present study is conspiring with the following objectives:

1. To determine the inhibition efficiency of *Musa sapientum* peel, *Emblica officinalis* and *Azadirachta indica* leaves extract.
2. To sound the formation of protective film on metal surface.

2. Corrosion Protection of Reinforced Concrete

The adulteration of reinforced concrete is principally due to the corrosion of the reinforcement, which occurs when the concrete has lost its ability to protect them. External pollutants found in air or water enters through the micro-pores in the concrete. In time, the chemical properties of the concrete in the vicinity of the reinforcement are altered and its pH drops. Once the ambient moisture penetrates through to the steel, corrosion begins. The prevention of corrosion can take numerous forms depending on the circumstances of the metal being corroded. The prevention techniques can be generally classified into following groups such as design improvement, metal selection, change of environment, electrochemical protection, coating and plating of metal and use of corrosion inhibitor.

3. Importance of Green Inhibitors:

The term corrosion is destruction of metal or alloy; they deteriorate slowly by the action of atmospheric gases, moisture and other chemicals. The rate at which it can propagate with a view of improving the lifetime of metallic and alloy materials. A corrosion inhibitor is a substance which, when added in small concentration to an environment, effectively reduce the corrosion rate of a metal exposed to that environment. To overcome those destruction of metal, different plant extract can be used as corrosion inhibitors, which is commonly known as green corrosion inhibitors. Plants are sources of naturally occurring compounds.

Naturally occurring compounds are mostly used because they are environmentally acceptable, cost effective and have abundant availability. Some with complex molecular structures and having different chemical, biological and physical properties. By adding inhibitors in low concentrations to corrosive media is to delay the reaction between the metal and the corrosive species in the medium.

The properties that must be met by corrosion inhibitor as:

- 1) Capability of reducing corrosion rate
- 2) The active principle of the corrosion inhibitor must be in contact with the metal
- 3) Must not have side effects.

4. Review of the Literature Survey:

Mild steel possess excellent ductility because of carbon contain, which is essential for the deep drawing of sheet but their strength is low. As the percentage is increased (0.2%) the strength steel rises into range required. Reduction the carbon percentage and addition small amount of other alloying elements can achieve this, mild steel possess the desirable properties as good tensile strength. With the above properties along with relative cheapness has made mild steel a desirable material for the use in industry for many mechanical and structural engineering purpose of bridge work, reactors, boiler plates, parts of various components and engines.

A great deal of study has been devoted to corrosion in the multidiscipline area of different. With the help of certain materials, including eco-friendly extract a number of organic compounds have been used as potent corrosion inhibitor.

There are several literature based on this topics were considered in order to formulate the project. There are many international journals as well as to national journals on this topic. The journals were studied and thus many corrosion problems were identified. Based on this topic some literatures were summarized in the following text.

Eddy, N. O And Ebenso, E. E (2008) investigate that the corrosion inhibition of mild steel by ethanol extract of *Musa sapientum* peels in H_2SO_4 has been studied using gasometric and thermometric methods. The results of the study reveals that the different concentrations of ethanol extract of *sapientum* peels inhibit mild steel corrosion. Values of activation energy of the inhibited corrosion reaction of mild steel are greater than the

value obtained for the blank. It is seen that the pH of the corrode decreases as the immersion time increases indicating enhancement in the rate of corrosion as the period of immersion increases. It is found that the inhibitive action of *Musa sapientum* peels is basically controlled by temperature, pH, and period of immersion, electrode potential and concentration of the inhibitor.

Salami, L., Wewe, T.O.Y, Akinyemi, O.P. And Patinvoh, R.J. (2009) Investigate the use of local wastes which are organic in nature for the production of green corrosion inhibitor is no doubt the trend of the day. Which evaluate to produce corrosion inhibitor from *Musa Sapientum* peels extract with a view of determining the effectiveness of the corrosion inhibitor? *Musa Sapientum* peels extract produced was used as a corrosion inhibitor on mild steel in concentrated tetraoxosulphate (VI) acid using weight loss method. The results of the study showed that as the concentration of the produced inhibitor increases, the rate of corrosion decreases. It also showed that as the concentration of the inhibitor increases, the inhibitor efficiency also increases up to an optimum of approximately 71 % for 0.8 g/l extract in 2.0M H₂SO₄ which is encouraging.

Int. J. Electrochem. Sci., (2011) to investigate the inhibitive action of eggplant peel extract toward the corrosion of mild steel in 2M HCl solution by employing weight loss measurements and electrochemical techniques. The results of both methods showed that the corrosion rate is decreasing and inhibition efficiency is increasing as the concentration of the eggplant inhibitor increased. The adsorption data were analyzed using various adsorption isotherm models and the results at temperatures of 25, 40 and 50°C have shown that the adsorption behaviour of eggplant extract molecules is best described by the Langmuir adsorption isotherm.

M. A. Quraishi (2011) that the strict environmental legislations and increasing ecological awareness among scientists have led to the development of “green” alternatives to mitigate corrosion. In the present work, literature on green corrosion inhibitors has been reviewed, and the salient features of our work on green corrosion inhibitors have been highlighted. Among the studied leaves, extract *Andrographis pediculate* showed better inhibition performance (98%) than the other leaves extract. *Strychnos nuxvomica* showed better inhibition (98%) than the other seed extracts. *Moringa oleifera* is reflected as a good corrosion inhibitor of mild steel in 1M HCl with 98% inhibition efficiency among the studied fruits extract. *Bacopa monnieri* showed its maximum inhibition performance to be 95% at 600 ppm among the investigated stem extracts. All the reported plant extracts were found to inhibit the corrosion of mild steel in acid media.

Int. J. Electrochem. Sci., (2012) Chemical constituents of purified sample of *Ficus platyphylla* gum was investigated using GCMS technique. *Ficus platyphylla* gum exudate has been found to be a good adsorption inhibitor for the corrosion of mild steel in solutions of HCL. The inhibition efficiency of the gum increased with increasing concentration but decreased with increase in temperature. Analysis of FTIR spectra revealed that *Ficus platyphyllagum* inhibited the corrosion of mild steel in solutions of HCl through the formation of inhibitor-metal complex (via C=O and NH₂ bonds) and adsorption through OH bond. The adsorption of the inhibitor was also found to be spontaneous, exothermic and supported the mechanism of physical adsorption.

M. Sangeetha et al., (2012) to replace the environmentally hazardous chromates, several non – chromates have been used as corrosion inhibitors. Extracts of plant materials top the list. The plant extracts are environmentally friendly, non- toxic and readily available. These extracts contain many ingredients. They contain several organic compounds which have polar atoms such as O, N, P and S. They are adsorbed onto the metal surface through these polar atoms; protective films are formed. Adsorptions of these ingredients obey various adsorption isotherms. The films have been analysed by many surface analysis techniques such as AFM, FTIR, UV, Fluorescence spectra and SEM.

N.Gunavathy.,S.C.Murugavel.,(2013) studied that the Inhibition efficiency of acid extract of dry *Musa acuminata* bract as corrosion inhibitor for mild steel in 1N hydrochloric acid was investigated in the present study. Experimental methods include weight loss, polarization and surface analysis studies. The inhibition efficiency increased with increase in inhibitor concentration and decreased with temperature suggesting the occurrence of physical adsorption. Potentiodynamic polarization curves reveal that inhibitor perform as a mixed type of inhibitor. The electrochemical impedance study showed a decrease in double layer capacitance as the adsorption of inhibitor lead to structural change at electrode-solution interface thereby controlling the mild steel dissolution by charge transfer mechanism. The mild steel surface in the absence and presence of the inhibitor was analysed by scanning electron microscopy (SEM). Corrosion inhibitor has

efficiency as high as 94.93% at 2% v/v inhibitor concentration. Inhibition mechanism can be attributed to the adsorption of phytochemical compounds of the bract extract on the surface of the mild steel.

Richard A.Ukpe, And Nnabuk Okon Eddy (2014) investigate the effectiveness of ethanol extract of mango peel waste for the inhibition of the corrosion of aluminum in solution of HCl using weight loss and gasometric methods. The results obtained indicated that ethanol extract of mango peel waste displayed average inhibition potential for the corrosion of aluminium in solution of HCl. The inhibited corrosion reaction supported a first order kinetic while its adsorption was exothermic, spontaneous and consistent with the mechanism of physisorption. The adsorption behaviour of the inhibitor was best described by the Langmuir, Temkin, El awardy et al and Dubinin-Raduskevich adsorption models.

Pereware Adowei Et Al (2014) has reported that the biomass of Nypa palm petiole and shoot have been investigated for their prospective use as adsorbent for elimination of different metal ions in aqueous solutions. Also ethanol extract of the fruit of Nypa fruticans has been studied for its use as a corrosion inhibitor of Zn in acidic solution. These studies indicate that the biomass of the different part of Nypa fruticans may possibly be employed for removal of toxic and valuable metals from aqueous systems, and could as well serve as a green corrosion inhibitor. The laboratory-scale batch sorption study of 2-(N,N-dimethyl-4-aminophenyl)-azo-benzene carboxylic acid (DMABA) in aqueous solution using carbonized and surface-modified carbon produced from Nypa palm leaves indicate that Nypa fruticans plant has considerable potential for the treatment of organic contaminants from wastewaters.

Janaina Cardozo Da Rocha et al., (2014) observed that the aqueous extracts of mango and orange peels were shown to be good corrosion inhibitors for carbon steel in a 1 M HCl solution. In the presence of 400 mg of mango and orange peel extracts, the weight loss measurements showed an increase in the inhibition efficiency with immersion time, where the best results after 24 h of immersion were 97% and 95%, respectively. The inhibitory effect was performed via the adsorption of compounds present in the fruit peel extracts on the steel surface. The inhibition efficiency obtained from this new extraction procedure, which contains more polar hetero-sides, is extremely similar to that obtained from a simple infusion. Thus, the more polar hetero-sides are likely responsible for the corrosion inhibition in both extracts. Due to the hydroxyl presence in these molecules, the adsorption on the metal surface could occur through H-bonding, which involves the displacement of water molecules from the metal surface.

A.O. Odiongenyi, (2015) investigate that the inhibition of the corrosion of zinc in solution of HCl was studied using gravimetric method. The results obtained indicated that ethanol extract of honey (EEH) exhibited minimum and maximum inhibition efficiencies of 30.57 and 72.01 % at 333 and 303 K, respectively. Generally, the inhibition efficiency of the extract was found to increase with increasing concentration but decreases with increase in temperature. From the range of values obtained for the activation energy, the pattern of variation of the inhibition efficiency of the extract with temperature and the range of values obtained for standard free energy of adsorption, the mechanism of physical adsorption was upheld for the adsorption of honey on mild steel surface.

Nitin Mathur And R .C. Chhipa (2015) investigate Corrosion of Mild steel IS: 432 (Part I)-1982 in various concentrations of hydrochloric acid and sulphuric acid are studied by Mass Loss method in the absence and presence of the extracts of seeds plant Pennisetum glaucum. Corrosion relates effects the technologies. Water plays an important role in enhancing corrosion. In this paper the mass loss equation, concludes that corrosion inhibition increased to proportionate concentration of the extract. It has been found that seed extract of plant Pennisetum glaucum is the effective and have high corrosion inhibition efficacy. It was also observed that mild steel is more susceptible in acid like HCl and H₂SO₄ solution.

Renita D'souza and Amit Chattree.,(2015) the inhibition potentials of Emblica officinalis leaves extract was investigated by evaluating the corrosion behaviour of mild steel immersed in 1 N HCL solution containing varied concentration of extract range from 200 ppm to 1000 ppm. The corrosion inhibitor of the extract was evaluated using weight loss method. the inhibition efficiency increased with increase in concentration of the inhibitor and the highest %IE was obtained at 1000 ppm. EOL extract formed a protective layer on the surface of the mild steel and improved surface condition, due to absorption, for the corrosion protection by UV and SEM analysis.

Based on the above literature, corrosion inhibitor were prepared with plant extract such as *Embllica officinalis* leaves, *Azadirachata indica* leaves and *musa sapientum* peel core at a different concentration of temperature and pH . Since mild steel is assumed to have corrosion nature due moisture present in atmosphere, green inhibitor helps to reduce the defects take place mild steel. Experimental study on

4. Conclusion

As it has been seen through this chapter, corrosion inhibitors are economically feasible to mitigate the problem caused by corrosion in the mild steel. A number of alternative plant extract are currently emerging oriented towards minimizing environmental impact providing effective corrosion inhibition. Collection of mild steel bar and various plant extract such as *musa sapientum* peel , *Embllica officinalis* leaves and *Azadirachta indica* leaf. Preparation of green corrosion inhibitors and immersion of mild steel bar into to solution 2M H_2SO_4 with and without inhibitors. Experimental test such as weight loss method, gasometric method and thermometric method for identify efficiency of corrosion rate.Finally analysis the surface study by various method such as SEM analysis and AFM analysis will be done and report. By analysing the corrosion rate of mild steel using various plant extract , highest %IE of plant extract was finalised and used in the concrete structure .

6.Paper Work

With the help of analysing the highest % IE of plant extract will finalize and used in the reinforcement concrete structure. Crowing the report with the help of comparing the concrete structure using and without using green inhibitor.

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